

Seat No.: _____

Enrolment No. _____

GUJARAT TECHNOLOGICAL UNIVERSITY

BE - SEMESTER-I (NEW) EXAMINATION – WINTER 2023

Subject Code: 3110001

Date: 12-01-2024

Subject Name: Chemistry

Time: 02:30 PM TO 05:00 PM

Total Marks: 70

Instructions:

1. Attempt all questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.
4. Simple and non-programmable scientific calculators are allowed.

		Marks
Q.1	(a) Define polar and nonpolar covalent bonds.	03
	(b) Differentiate between temporary and permanent hardness.	04
	(c) Explain Lime soda process with its advantages and disadvantages.	07
Q.2	(a) Discuss four important physical properties of metals.	03
	(b) Explain the shapes of s, p, d and f orbitals.	04
	(c) How will you protect metal from corrosion?? Explain wet corrosion mechanism.	07
	OR	
	(c) Explain hard soft acids and bases with their properties.	07
Q.3	(a) What is nano material? Explain the term with historical aspect.	03
	(b) Write a short note on 'top down' and 'bottom-up approach' to synthesize Nanomaterials.	04
	(c) Define polymerization. Give major classification of polymers with suitable examples and applications.	07
	OR	
Q.3	(a) Define Stretching and bending vibration in IR spectroscopy.	03
	(b) Define pH. Calculate pH of 0.1 N HCl.	04
	(c) Explain the preparation, properties and any one specific use of the Nylon-6,6 and Buna-S rubber.	07
Q.4	(a) State important applications of nanomaterial in catalysis.	03
	(b) Discuss major disadvantages of using hard water in various industries.	04
	(c) Explain the IR spectroscopy with its application.	07
	OR	
Q.4	(a) What is a biofertilizer?	03
	(b) Explain Importance of Biofuels.	04
	(c) Define Fermentation. Explain manufacturing of Acetic acid by fermentation process.	07
Q.5	(a) Define the following terms: Octane and Cetane Number	03
	(b) Discuss the Selection rules for UV-Visible spectroscopy.	04
	(c) What are bioreactors? Discuss the applications of biotechnology.	07

OR

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| Q.5 | (a) | Explain Liquid Crystals. | 03 |
| | (b) | Write characteristics of a good fuel. | 04 |
| | (c) | Give classification of fuel. Discuss proximate analysis of fuel. | 07 |
