

GUJARAT TECHNOLOGICAL UNIVERSITY
B.PHARM – SEMESTER –4 EXAMINATION – SUMMER-2024

Subject Code: BP403TP**Date: 12/06/2024****Subject Name: Physical Pharmaceutics II****Time: 10.30 a.m. to 1.30 p.m.****Total Marks: 80****Instructions:**

1. Attempt any five questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.

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| Q.1 | (a) Classify the different types of colloids in brief. | 06 |
| | (b) Write note on types of emulsion. | 05 |
| | (c) Describe the DLVO theory with energy curve. | 05 |
| Q.2 | (a) Explain cup and bob viscometer with a labelled diagram. | 06 |
| | (b) Differentiate Newtonian and Non-Newtonian systems with suitable examples. | 05 |
| | (c) Define: Kinematic viscosity, Thixotropy, Spur, Rheopexy, Fluidity. | 05 |
| Q.3 | (a) Describe physical instability markers of emulsion. | 06 |
| | (b) Explain the sedimentation volume and degree of flocculation of suspension. | 05 |
| | (c) Differentiate flocculated suspension and deflocculated suspension. | 05 |
| Q.4 | (a) Enlist the methods for determination of particle size and explain conductivity method in detail. | 06 |
| | (b) Enumerate the derived properties of powders and explain angle of repose. | 05 |
| | (c) What is specific surface of particles? Discuss air permeability method in brief. | 05 |
| Q.5 | (a) Enlist the factors affecting rate of reaction and explain temperature in detail. | 06 |
| | (b) Discuss photolytic degradation and its prevention. | 05 |
| | (c) Derive the first order rate of reaction equation. | 05 |
| Q. 6 | (a) Discuss the optical properties of colloids in brief. | 06 |
| | (b) Describe the Heckle equation. | 05 |
| | (c) Define: Stoke's diameter, Projected diameter, Surface diameter, Micromeritics, Carr's index. | 05 |
| Q.7 | (a) Write a note on accelerated stability study with its limitations. | 06 |
| | (b) Give note on negative Thixotropy. | 05 |
| | (c) Comment: Emulsions are thermodynamically unstable. Justify. Define colloids and explain the term Gold number. | 05 |
